

Answers - Point Source Lab

Calculation 1. Boiler Emissions (CEM Data)

Calculating Hourly Emissions of SO₂

$$E_{\text{SO}_2} = \frac{(C \times \text{MW} \times Q \times 60)}{(V \times 10^6)}$$

$$E_{\text{SO}_2} = \frac{1,058.8 \times 64 \times 155,272 \times 60}{385.5 \times 10^6} = 1,637 \text{ lb/hr}$$

Calculation 1. Boiler Emissions (CEM Data)

Calculating Heat Input

$$H_{in} = \frac{(Q_f \times HHV)}{(10^6)}$$

$$H_{in} = \frac{46,000 \times 18,000}{10^6} = 828 \text{ MMBtu/hr}$$

Calculation 1. Boiler Emissions (CEM Data)

Developing SO₂ Emission Factors

$$EF_{SO_2} = \frac{E_{SO_2}}{H_{in}} = \frac{1,637 \text{ lb/hr}}{828 \text{ MMBtu/hr}} = 1.98 \text{ lb/MMBtu}$$

Calculation 1. Boiler Emissions (CEM Data)

Calculating Annual SO₂ Emissions

Annual SO₂ Emissions = Hourly SO₂ emissions
x Op Hrs

$$= \frac{(1,637 \text{ lb/hr} \times 5,400 \text{ hrs})}{(2,000 \text{ lb/ton})} = 4,419 \text{ tons per year}$$

Calculation 2. Boiler Emissions (Source Testing)

$$P_b \text{ Emission Rate} = P_b \text{ concentration} \\ \times \text{stack gas flow rate}$$

$$= 0.0005 \text{ lb/dscf} \times 51,700 \text{ dscf/min} \times 60 \text{ min/hr} \\ = 1,551 \text{ lb/hr}$$

$$\text{Annual } P_b \text{ Emissions} = 1,551 \text{ lb/hr} \times 5,840 \text{ hr/yr} \\ \times 1 \text{ ton}/2,000 \text{ lb} = 4,528 \text{ tpy}$$

Calculation 3. Copper Coil Manufacturing (Mass Balance)

Consumption = (mass of coil and oil to
annealer - mass of coil processed) x 85
percent

$$= (5,075 \text{ kg} - 5,000 \text{ kg}) \times 0.85$$

$$= 64 \text{ kg oil destroyed in the annealer}$$

Calculation 3. Copper Coil Manufacturing (Mass Balance)

Estimate of Output (emissions)

Output = Input - Consumption - Accumulation

= 3,000 kg - 64 kg - 2,800 kg

= 136 kg

The VOC emissions associated with this process are thus 136 kg oil pr 5,000 kg of copper coil process, or 0.027 kg oil per kg of copper coil processed.

Calculation 4. Boiler Emissions (Fuel Analysis)

$$E_{\text{SO}_2} = Q_f \times \text{pollutant concentration in fuel} \times \left(\frac{\text{MW}_p}{\text{MW}_f} \right)$$

$$= 5,000 \text{ lb/hr} \times 0.01 \times (64/32)$$

$$= 100 \text{ lb/hr}$$

$$= 100 \text{ lb/hr} \times 8,760 \text{ hr/yr} \times \frac{1 \text{ ton}}{2,000 \text{ lbs}}$$

$$= 438 \text{ tons/year of SO}_2$$

Calculation 5. Coal-fired Industrial Boiler (Emission Factors and Temporal Allocation)

Boiler Emissions = Annual Coal Consumption x
Emission Factor

$$\text{TOC} = 928,000 \text{ tons/year} \times 0.3 \text{ lb/ton} = 278,400 \text{ lb/year} \\ = 139.2 \text{ tpy}$$

$$\text{P}_b = 928,000 \text{ tons/year} \times 8.9\text{E-}03 \text{ lb/ton} = 8,259 \text{ lb/year} \\ = 4.1 \text{ tpy}$$

$$\text{NO}_x = 928,000 \text{ tons/year} \times 9 \text{ lb/ton} = 8,352,000 \text{ lb/year} \\ = 4,176 \text{ tpy}$$

$$\text{CO} = 928,000 \text{ tons/year} \times 0.6 \text{ lb/ton} = 556,800 \text{ lb/year} \\ = 278 \text{ tpy}$$

Calculation 5. Coal-fired Industrial Boiler (Emission Factors and Temporal Allocation)

Estimating Uncontrolled Emissions for SO₂

$$\text{SO}_2 \text{ Emissions} = \text{Annual Coal Consumption} \times (\text{Emission Factor} \times \text{Coal Sulfur Content})$$

$$= 928,000 \text{ tons/year} \times (39 \text{ lb/ton} \times 1.87) = 67,679,040 \text{ lb/year}$$

$$= 33,840 \text{ tpy}$$

Calculation 5. Coal-fired Industrial Boiler (Emission Factors and Temporal Allocation)

Estimating Controlled Emissions

$$\text{Controlled Emissions} = \text{Uncontrolled Emissions} \times (1 - \text{Efficiency}/100)$$

$$\text{Total PM} = 2,858 \text{ tpy} \times (1 - 75/100) = 2,858 \text{ tpy} \times (0.25) = 715 \text{ tpy}$$

$$\text{SO}_2 = 33,840 \text{ tpy} \times (1 - 93/100) = 33,840 \text{ tpy} \times (0.07) = 2,369 \text{ tpy}$$

Calculation 5. Coal-fired Industrial Boiler (Emission Factors and Temporal Allocation)

Temporal Allocation

Seasonal emissions = Seasonal throughput
fraction x annual emissions

Winter PM = $0.5 \times 715 \text{ tpy} = 357.5 \text{ tons}$

Spring PM = $0.2 \times 715 \text{ tpy} = 143 \text{ tons}$

Summer PM = $0.1 \times 715 \text{ tpy} = 71.5 \text{ tons}$

Fall PM = $0.2 \times 715 \text{ tpy} = 143 \text{ tons}$

Answers - Area Source Lab

Calculation 1. Estimating County Level Wood Usage (Top-Down Approach)

County Wood Use = State wood use x county
households/state households

= 622,000 x 1,242/80,047 = 9,651 cords
burned in county of study

Calculation 1. Estimating County Level Wood Usage (Top-Down Approach)

Total waste generated = Total number of
households using burn barrels x average waste
generated per household

$$= 997 \times 6.75 = 6,730 \text{ lb/day}$$

Total Combustible waste generated by households
= Total waste generated x percentage of
combustible waste

$$= 6,730 \text{ lb/day} \times 80/100 = 5,384 \text{ lb/day}$$

Total combustible waste = 5,384 lb/day = 2.69
ton/day

Calculation 1. Estimating County Level Wood Usage (Top-Down Approach)

Daily Emissions = Emission Factor

x Total Combustible waste generated (in tons)

$$\text{CO} = 85 \text{ lb/ton} \times 2.69 \text{ tons/day} = 228.6 \text{ lb/day}$$

$$\text{PM} = 16 \text{ lb/ton} \times 2.69 \text{ tons/day} = 43.03 \text{ lb/day}$$

$$\text{SO}_x = 1 \text{ lb/ton} \times 2.69 \text{ tons/day} = 2.69 \text{ lb/day}$$

$$\text{NO}_x = 6 \text{ lb/ton} \times 2.69 \text{ tons/day} = 16.14 \text{ lb/day}$$

$$\text{TOC} = 21.5 \text{ lb/ton} \times 2.69 \text{ tons/day} = 57.84 \text{ lb/day}$$

TOC = Total organic compound

Calculation 3. State I Gasoline Marketing (Rule Effectiveness/Rule Penetration)

Calculating Emissions

$$\begin{aligned} E &= A \times EF \times 1 - (CE \times RE \times RP) \\ &= 500 \times 11.5 \times 1 - ((0.95) (0.8) (0.93)) \\ &= 1,685 \text{ lb of VOC/day} \end{aligned}$$

Calculation 4. Surface Coating Operations (Mass Balances)

Estimating Uncontrolled PM Emissions

Coating density of solid content = Density of
coating A - Density of volatile content

$$= 7.5 \text{ lb/gal} - 6.2 \text{ lb/gal} = 1.3 \text{ lb solids/gal}$$

Uncontrolled PM emissions = Density of solid
content x annual usage x (1-transfer efficiency)

$$= 1.3 \text{ lb/gal} \times 1,600 \text{ gal/yr} \times (1 - 0.6)$$

$$= 832 \text{ lb/yr or } 0.416 \text{ tpy}$$

Calculation 4. Surface Coating Operations (Mass Balances)

Estimating Controlled PM Emissions

Controlled PM Emissions = Uncontrolled PM
x (1 - Control efficiency)

$$= 0.416 \times (1 - 99/100) = 0.00416 \text{ tpy of solids}$$

Calculation 5. Surface Coating (Per Employee Emission Factor)

Calculating Area Source Emissions

VOC emissions from area sources = area
source employment x average coating usage
x % VOC x coating density

$$= 900 \text{ employees} \times 12 \text{ gal/yr} \times 7.5 \text{ lb/gal} \times \frac{45}{100}$$

$$= 36,450 \text{ lb/yr}$$

$$= 18.23 \text{ tpy of VOC}$$

Calculation 6. Emissions from Benzene Loading Operations (Emission Factor)

$$L_L = 12.46 \times \frac{S \times P \times M}{T}$$

$$L_L \text{ (lb/103 gal)} = \frac{12.46 \times 0.6 \times 2 \text{ psia} \times 78 \frac{\text{lb}}{\text{lb - Mole}}}{540^\circ \text{ R}}$$

$$= 2.16 \text{ lb benzene/1,000 gal}$$

Toxics Lab

Example Calculation #1

Actual Annual and Potential Acetaldehyde Uncontrolled Emissions from a Natural Gas-Fired Heater

Given: Actual hours of operation = 2400 hr/yr at 80% of maximum capacity

Heat content = 1050 MMBtu/MMCF

Heat input rate = 2 MMBtu/hr

Acetaldehyde emission factor for commercial-sized heater (see NTI documentation) = 1.3 E-08 lb acetaldehyde/MM Btu

Example Calculation #1 (Continued)

$$\text{Actual Acet emissions} \left(\frac{\text{lb}}{\text{yr}} \right) = (EF)(\text{heat input rate})(\text{annual operation})(\% \text{ firing capacity})$$

where:

EF = pollutant emission factor in lb/MMBtu and

heat input rate = maximum firing rate in MMBtu/hr.

Example Calculation #1

(Continued)

$$\text{Actual Emissions} = (1.3 \times 10^{-8} \frac{\text{lb}}{\text{MMBtu}})(2 \frac{\text{MMBtu}}{\text{hr}})(2400 \frac{\text{hr}}{\text{yr}})(80\%)$$

$$\text{Potential Emissions} = (1.3 \times 10^{-8} \frac{\text{lb}}{\text{MMBtu}})(2 \frac{\text{MMBtu}}{\text{hr}})(2400 \frac{\text{hr}}{\text{yr}})(100\%)$$

$$\text{Actual Emissions} = 4.99 \times 10^{-5} \text{ lb/yr}$$

$$\text{Potential Emissions} = 6.24 \times 10^{-5} \text{ lb/yr}$$

Example Calculation #2

Uncontrolled Actual Annual and Potential Xylene Emissions from a Silk Screen Printing Process

Given:

Constituents of ink thinner (100% VOC by weight, 5% xylene by weight)

Usage of ink thinner = 45 lb/yr

Constituents of ink (10% VOC by weight, 2% xylene by weight)

Usage of ink = 204 lb/yr

Estimated maximum hourly usage = 1 lb ink and 0.3 lb thinner

Example Calculation #2 (Continued)

$$\begin{aligned}\text{Actual xylene emissions } \left(\frac{\text{lb}}{\text{yr}} \right) &= \left[\left(\left(45 \frac{\text{lb}}{\text{yr}} \right) \left(0.05 \frac{\text{lb xylene}}{\text{lb}} \right) \right) + \left(\left(204 \frac{\text{lb}}{\text{yr}} \right) \left(0.02 \frac{\text{lb xylene}}{\text{lb}} \right) \right) \right] \\ &= 6.33 \frac{\text{lb xylene}}{\text{yr}}\end{aligned}$$

$$\begin{aligned}\text{Potential xylene emissions } \left(\frac{\text{lb}}{\text{yr}} \right) &= \left[\left(\left(0.3 \frac{\text{lb}}{\text{hr}} \right) \left(0.05 \frac{\text{lb xylene}}{\text{lb}} \right) \right) + \left(\left(1 \frac{\text{lb}}{\text{hr}} \right) \left(0.02 \frac{\text{lb xylene}}{\text{lb}} \right) \right) \right] * 8760 \frac{\text{hr}}{\text{yr}} \\ &= 307 \frac{\text{lb xylene}}{\text{yr}}\end{aligned}$$

Example Calculation #3

Cr(VI) Emissions from a Surface Coating Operation

Given: Coating usage = 500 gal/year

Cr(VI) concentration = 0.08% Cr(VI), by mass

Density = 8 lb/gal

Coating transfer efficiency = 40%

Air pollution control device = 80%

Example Calculation #3 (Continued)

$$E_x = (Q_{in} - Q_{out}) * C_x$$

Where: E_x = Total emissions of pollutant x
 Q_{in} = Quantity of material entering the process
 Q_{out} = Quantity of material leaving the process as waste, recovered,
or in product
 C_x = Concentration of pollutant x

Example Calculation #3 (Continued)

Mass of Cr(VI) entering spray booth:

$$\begin{aligned} &= 500 \text{ gal/yr} \times 8 \text{ lb/gal} \times .0008 \text{ lb Cr(VI)/lb coating} \\ &= 3.2 \text{ lb/yr} \end{aligned}$$

Mass of Cr(VI) removed from spray booth:

From transfer to surface

$$3.2 \times 0.4 = 1.28 \text{ lbs/yr}$$

Leaving 1.92 lb/yr

From air pollution control device

$$1.92 \times 0.8 = 1.54 \text{ lb/yr}$$

Example Calculation #3 (Continued)

$$\begin{aligned} \text{Emission rate} &= 3.2 \text{ lb/yr} - 1.28 \text{ lb/yr} - 1.54 \text{ lb/yr} \\ &= 0.38 \text{ lb/yr} \end{aligned}$$

or

$$\begin{aligned} \text{Emission rate} &= 3.2 \text{ lb/yr} \times (1-0.4) \times (1-0.8) \\ &= 0.38 \text{ lb/yr} \end{aligned}$$